

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

STEREOSPECIFIC ONE-POT SYNTHESIS OF 5-CARBOXY-3,4-DIPHENYLIMIDAZOLIDIN-2-ONE, A USEFUL PRECURSOR TO THE CORRESPONDING 2,3-DIAMINOPROPANOIC ACID

Ivanka K. Kavrakova^a; Maria J. Lyapova^a

^a Institute of Organic Chemistry with Centre of Phytochemistry Bulgarian Academy of Sciences, Sofia, BULGARIA

To cite this Article Kavrakova, Ivanka K. and Lyapova, Maria J.(1996) 'STEREOSPECIFIC ONE-POT SYNTHESIS OF 5-CARBOXY-3,4-DIPHENYLIMIDAZOLIDIN-2-ONE, A USEFUL PRECURSOR TO THE CORRESPONDING 2,3-DIAMINOPROPANOIC ACID', *Organic Preparations and Procedures International*, 28: 3, 333 – 338

To link to this Article: DOI: 10.1080/00304949609356539

URL: <http://dx.doi.org/10.1080/00304949609356539>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

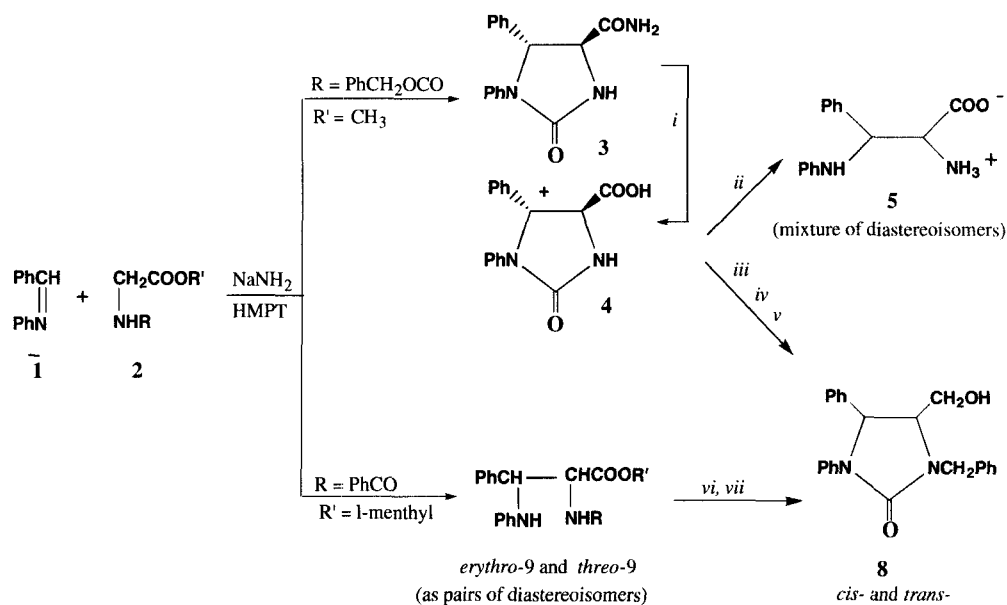
**STEREOSPECIFIC ONE-POT SYNTHESIS OF
5-CARBOXY-3,4-DIPHENYLIMIDAZOLIDIN-2-ONE, A USEFUL PRECURSOR
TO THE CORRESPONDING 2,3-DIAMINOPROPANOIC ACID**

Ivanka K. Kavrakova and Maria J. Lyapova*

*Institute of Organic Chemistry with Centre of Phytochemistry
Bulgarian Academy of Sciences, 1113 Sofia, BULGARIA*

Recently we reported on the synthesis of 2-benzoylamino-3-phenyl-3-phenylamino-propanoic acid menthyl esters and the corresponding chiral diamino propanols¹ which are of interest as tridentate auxiliaries in asymmetric synthesis. They were readily obtained in a two-step procedure starting from *N*-benzylideneaniline and menthyl hippurate in the presence of sodium amide. The high yield and diastereoselectivity of this reaction prompted us to study the aldol-like condensation of *N*-benzylideneaniline with other glycine equivalents such as *N*-carbobenzyglycine methyl ester **2** as the CH-acidic component. We expected this reaction to provide an easy access to imidazolidine derivatives similar to the previously reported² aldol-type reaction between benzaldehyde and *N*-carbobenzyglycine methyl ester leading to the corresponding *trans*-oxazolidin-2-one. Imidazolidine derivatives are usually products of multi-step reactions and are desirable precursors for the preparation of 2,3-diamino acids.³ The latter are uncommon, naturally occurring amino acids that have recently attracted considerable interest. We now report the stereospecific one-pot synthesis of *trans*-5-carboxy-3,4-diphenylimidazolidin-2-one (**4**) and its further hydrolysis to the corresponding 2-amino-3-phenyl-3-phenylaminopropanoic acid (**5**).

When the dianion of **2** obtained from the reaction with two equivalents of sodium amide was allowed to react with *N*-benzylideneaniline in hexamethylphosphoramide (HMPT), compounds **3** and **4** were obtained as single diastereoisomers. The formation of carboxamide **3** is probably due to partial ammonolysis of the initially formed methoxycarbonylimidazolidin-2-one. When heated with dilute sulfuric acid, the carboxamide **3** chemoselectively gave the cyclic acid **4** as a sole product. The imidazolidin-2-one ring in compounds **3** and **4** proved to be highly resistant to both acidic and alkaline hydrolysis under various conditions. Obviously the 3,4-disubstitution makes the hydrolysis much more difficult, as 3- or 4- monosubstituted 5-carboxylimidazolidin-2-ones afforded the open-chain acids when heated with conc. hydrochloric acid.^{3a,3b} The 2-amino-3-phenyl-3-phenylaminopropanoic acid **5** was obtained in a 40% yield as a (1:1.2) diastereoisomeric mixture only on treatment of **4** with 4*N* NaOH at 110 for 6 days. We assigned the configuration of acid **4** by correlating it with the two



Reagents: *i*) 20% H₂SO₄, heat; *ii*.) 4N NaOH, heat; *iii*) MeOH, BF₃/Et₂O, heat; *iv*.) BzBr/Ag₂O, DMF; *v*) NaBH₄, MeOH; *vi*) LiAlH₄, Et₂O; *vii*) COCl₂, 10% KOH, toluene.

isomeric imidazolidin-2-one derivatives **8** (Scheme) through a series of transformations not affecting the two chiral centres. For their synthesis, we employed our previously reported preparation of the four diastereoisomeric 2-benzoylamino-3-phenyl-3-phenylaminopropanoic acid menthyl esters.¹ The major, slower eluting pair of *threo* diastereoisomers **9a** as well as the faster eluting, minor *erythro* pair **9b** were separated by preparative HPLC. They were further transformed first to the racemic alcohols **10a** and **10b**, then to the corresponding diastereoisomeric imidazolidin-2-ones **8a** and **8b**. The configuration of the cyclic compounds **8** was established on the basis of parallel quantitative NOE measurements. A 15% enhancement of the signal of the 5-H proton upon irradiation of 4-H was observed for **8b**, indicating its *cis* configuration and a 3% enhancement for the *trans* isomer **8a**. From the configuration of the cyclic products **8** thus established, the *threo* configurations of the related open-chain compounds **9a** and **10a** and the *erythro* configurations of **9b** and **10b** follows. The imidazolidinocarboxylic acid **4** and the carboxamide **3**, being related with *trans*-**8**, also proved to have a *trans* configuration.

Although the yield is not high, this one-pot procedure for the synthesis of 3,4-diphenyl-imidazolidin-2-one derivatives of *trans* configuration, starting from easily available materials, is promising. These compounds could further be useful as reagents for the preparation of α -branched 2,3-diamino acids and as chiral auxiliaries in asymmetric synthesis when they are optically active.

EXPERIMENTAL SECTION

Mps. were determined on a Kofler microscope. IR spectra were recorded on a Specord 71 IR Karl Zeiss Iena spectrometer in CHCl₃. ¹H NMR spectra were obtained on a Bruker WM 250 MHz spec-

trometer in CDCl_3 , using SiMe_4 as an internal standard. Mass spectra were determined on a Jeol IMSD 300 mass spectrometer at 70 eV. Preparative HPLC was performed on an HPLC Perkin-Elmer 2/2 chromatograph: stationary phase, Si 100-7; mobile phase, hexane-diethyl ether (85:15 v/v); , 254 nm; flow rate, 6 mL/min⁻¹. IR, ¹H NMR and mass spectral data of products are given in Table.

***trans*-5-Carboxamide-3,4-diphenylimidazolidin-2-one (3) and *trans*-5-Carboxy-3,4-diphenylimidazolidin-2-one (4).**- To a stirred solution of *N*-benzylideneaniline (1) (2.5 mmol) and *N*-carboboxyglycine methyl ester⁴ (2) (2.5 mmol) in HMPT (1 mL) was added sodium amide (5.0 mmol). The reaction mixture was stirred for 1 hr at room temperature, hydrolyzed with water and extracted with ethyl acetate. The ethyl acetate solution was dried (MgSO_4) and evaporated to dryness under reduced pressure. ¹H NMR of the residue showed the presence of *trans*-5-carboxamide-3,4-diphenylimidazolidin-2-one 3 (a single diastereoisomer) and *N*-benzylideneaniline in a ratio 1:1.5. Crystallization from ethanol afforded pure 3 in 22% yield, mp. 230-232°. An analytically pure sample had mp. 231-232°.

Anal. Calcd. for $\text{C}_{16}\text{H}_{15}\text{N}_3\text{O}_2$: C, 68.31; H, 5.37; N, 14.94. Found: C, 68.44; H, 5.54; N, 15.17

The alkaline solution was acidified with 5N HCl. ¹H NMR of the ethyl acetate extract of an aliquot sample from the solution showed the presence of 4 (a single diastereoisomer) and *N*-carboboxyglycine in a ratio 1:1. The precipitate was filtered off and washed with water. 4 was obtained in 29% yield, mp. 93-96°. An analytically pure sample had mp. 95-98° (aq. ethanol).

Anal. Calcd. for $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_3$: C, 68.13; H, 5.00; N, 9.92. Found: C, 68.13; H, 5.21; N, 9.98

A stirred suspension of 3 (0.35 mmol) in 5mL 20% H_2SO_4 was heated at 100° for 2 hrs. The cooled mixture was diluted with water, the product was filtered off and washed with water. 4 was isolated in 99% yield, mp. 93-96°, identical with the obtained above.

2-Amino-3-phenyl-3-phenylaminopropanoic Acid (5).- The imidazolidinecarboxylic acid 4 (1.4 mmol) in 10mL 4N NaOH was heated for 6 days at 110° in a stoppered teflon tube. The solution was acidified with 5N HCl and extracted with methylene chloride. The acidic water solution was evaporated to dryness and the residue was extracted at boiling with dry ethanol (3 x 40mL). The solvent was evaporated under reduced pressure, the residual solid dissolved in 5mL of water and absorbed onto a strongly acidic cation exchanger. The column was washed with water to neutral reaction, then the aminoacid 5 was eluted with 12.5% aqueous NH_3 . Ninhydrin-positive fractions were combined and concentrated to dryness *in vacuo*. Compound 5 was obtained in 40% yield, mp. >230° (dec.).

Anal. Calcd. for $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}_2$: C, 70.29; H, 6.29; N, 10.93. Found: C, 70.69; H, 6.85; N, 10.59

***trans*-5-Methoxycarbonyl-3,4-diphenylimidazolidin-2-one (6).**- A solution of 4 (1.4 mmol) and BF_3 -etherate (3.3 mmol) in dry methanol (8 mL) was heated under reflux for 1 hr. The solvent was evaporated, the residue was treated with ice-water and extracted with methylene chloride. The extract was washed with 1M aqueous sodium hydrogencarbonate solution and water and dried (MgSO_4). After removal of the solvent *in vacuo* and trituration of the residue with pentane 6 was obtained in 93% yield, mp. 122-124°. An analytically pure sample had mp. 122-124° (aq. ethanol).

Anal. Calcd. for $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}_3$: C, 68.91; H, 5.44; N, 9.45. Found: C, 68.88; H, 5.41; N, 9.42

TABLE. Spectral Data of Products

Product	IR (cm ⁻¹)	¹ H NMR (δ) ^a	<i>m/z</i>
3	1667, 1693, 3126, 3247, 3353, 3359 ^{b,c}	3.78 (d, 1H, 5-H, J 3.9), 5.34 (d, 1H, 4-H, J 3.9), 6.90-7.51(m, 10H _{arom.}), 7.66 (s, 1H, NH, D ₂ O exch.). ^d	281 (M ⁺ , 46%), 237 (100), 194 (60), 182 (40), 118 (48), 104 (19), 91 (52), 77 (14).
4	1650, 1720, 3425	3.57 (br.s, 1H, NH, D ₂ O exch.), 4.02 (d, 1H, 5-H, J 3.5), 5.56 (d, 1H, 4-H, J 3.5), 5.57 (s, 1H, COOH, D ₂ O exch.), 6.95-7.45 (m, 10H _{arom.}).	282 (M ⁺ , 100%), 237 (49), 94 (47), 181 (51), 118 (30), 191 (47), 77 (55).
5	1450, 1540, 165, 3, 3434 (br.) ^{b,c}	3.67 (d, 1H, CH, J 4.3, diastereoisomer 1), 3.95 (d, 1H, CH, J 8.9, diastereoisomer 2), 4.04 (d, 1H, CH, J 4.3, diastereoisomer 1), 4.18 (d, 1H, CH, J 8.9, diastereoisomer 2), 5.95-6.32 (m, 10H _{arom.}). ^e	256 (M ⁺ , 2%), 221 (2), 210 (2), 182 (100), 104 (21), 76 (52).
6	1710, 1745, 3440	3.88 (s, 3H, CH ₃), 4.09 (d, 1H, 5-H, J 3.4), 5.49 (d, 1H, 4-H, J 3.5), 5.52 (s, 1H, NH, D ₂ O exch.), 6.99-7.46 (m, 10H _{arom.}).	296 (M ⁺ , 100%), 253 (6), 237 (89), 194 (60), 182 (47), 118 (40), 91 (30), 77 (28).
7	1705, 1745	3.75 (s, 3H, CH ₃), 3.79 (d, 1H, 5-H, J 3.7), 4.24 (d, 1H, CH ₂ Ph, J 15.2), 5.06 (d, 1H, CH ₂ Ph, J 15.2), 5.24 (d, 1H, 4-H, J 3.7), 6.96-7.47 (m, 15H _{arom.}).	386 (M ⁺ , 26%), 327 (61), 194 (4), 91 (100), 77 (7).
<i>cis</i> - 8	1702, 3587 ^b	1.56 (s, 1H, OH, D ₂ O exch.), 3.27-3.43 (m, 2H, CH ₂ OH), 3.88-3.96 (m, 1H, 5-H), 4.34 (d, 1H, CH ₂ Ph, J 15.3), 4.90 (d, 1H, CH ₂ Ph, J 15.3), 5.31 (d, 1H, 4-H, J 9.0), 6.93-7.47 (m, 15H _{arom.}).	358 (M ⁺ , 10%), 327 (54), 194 (4), 104 (4), 91 (100), 77 (8).
<i>trans</i> - 8	1695, 3550	2.21 (br.s, 1H, OH, D ₂ O exch.), 3.9-3.34 (m, 1H, 5-H), 3.53-3.77 (m, 2H, CH ₂ OH), 4.36 (d, 1H, CH ₂ Ph, J 15.3), 4.66 (d, 1H, CH ₂ Ph, J 15.3), 5.17 (d, 1H, 4-H, J 6.3), 6.95-7.49 (m, 15H _{arom.}).	358 (M ⁺ , 10%), 327 (50), 194 (4), 104 (8), 91 (100), 77 (15).
<i>erythro</i> - 10	3348, 3411 ^b	1.95 (s, 3H, NHPh, NHCH ₂ Ph, OH, D ₂ O exch.), 2.99 (dd, 1H, 2-H, J 4.8/9.6), 3.48-3.63 (m, 2H, CH ₂ OH), 3.85 (s, 2H, CH ₂ Ph), 4.62 (d, 1H, 3-H, J 5.1), 6.49-7.35 (m, 15H _{arom.}).	332 (M ⁺ , 1%), 181 (57), 150 (100), 104 (9), 91 (68), 77 (15).
<i>threo</i> - 10	3377, 3397 ^b	2.12 br.s, 2H, NHCH ₂ Ph, OH, D ₂ O exch.), 2.89-2.95 (m, 1H, 2-H), 3.44-3.79 (m, 4H, CH ₂ Ph, CH ₂ OH), 4.48 (d, 1H, 3-H, J 6.3), 4.78 (br.s, 1H, NHPh, D ₂ O exch.), 6.51-7.41 (m, 15H _{arom.}).	332 (M ⁺ , 1%), 181 (34), 150 (100), 104 (6), 91(53), 77(11).

a) J in Hz. b) Obtained on Bruker ISF 193V. c) KBr disc. d) In [²H₆]Me₂SO. e) In ²HCl.

trans-1-Benzyl-5-methoxycarbonyl-3,4-diphenylimidazolidin-2-one (7).- To a stirred solution of **6** (0.7 mmol) in DMF (1 mL) were added Ag₂O (0.9 mmol) and benzylbromide (2.0 mmol). The mixture was stirred at room temperature for 5 hrs, then evaporated *in vacuo*. The residue was dissolved in chloroform, the solution was washed with water, dried (MgSO₄) and evaporated to dryness under reduced pressure. Crystallization from isopropyl ether-pentane afforded **7** in 70% yield, mp. 92-95°. An analytically pure sample had mp. 96-98° (benzene-hexane).

Anal. Calcd. for C₂₄H₂₂N₂O₃: C, 74.59; H, 5.74; N, 7.25. Found: C, 74.45; H, 5.54; N, 7.25

threo- and erythro- 2-Benzoylamino-3-phenyl-3-phenylaminopropanoic acid Menthyl Esters (9a and 9b) were separated by preparative HPLC from the crude mixture of diastereoisomers obtained acc. to 1) (*t*/_s 279 and 359 for *threo-9* and 153 and 171 for *erythro-9*).

threo- and erythro- 2-Benzoylamino-3-phenyl-3-phenylaminopropan-1-ols (10a and 10b).- A suspension of *threo-9* (1 mmol) and LiAlH₄ (10 mmol) in diethyl ether (10 mL) was stirred for 6 hrs at room temperature and then hydrolyzed with water. The dried (MgSO₄) solution was evaporated and the crude product was steam-distilled on a rotary evaporator for separation of the menthol formed. The residue was dissolved in chloroform and dried (MgSO₄). Evaporation of the solvent and recrystallization from ethanol afforded a 63% yield of *threo-10*, mp. 105-106°.

Anal. Calcd. for C₂₂H₂₄N₂O: C, 79.48; H, 7.28; N, 8.43. Found: C, 79.29; H, 7.40; N, 8.65

Erythro-10 was obtained analogously from 0.2 mmol of the crude *erythro-9* in 49% yield, mp. 127-129°. An analytically pure sample had mp. 130-132° (ethanol).

Anal. Calcd. for C₂₂H₂₄N₂O: C, 79.48; H, 7.28; N, 8.43. Found: C, 79.62; H, 7.42; N, 8.63

trans-1-Benzyl-5-hydroxymethyl-3,4-diphenylimidazolidin-2-one (8a).- (a) To a mixture of *threo-10* (0.5 mmol) in benzene (5 mL) and 10% potassium hydroxide (3 mL), a toluene solution of phosgene (3 mL) was added in portions at 0° and the emulsion was stirred at 0° for 4 hrs. The organic layer was separated, washed with water and dried (MgSO₄). Evaporation and recrystallization from benzene-hexane afforded *trans-8* in 55% yield, mp. 153-155°.

Anal. Calcd. for C₂₃H₂₂N₂O₂: C, 77.07; H, 6.19; N, 7.82. Found: C, 77.12; H, 6.24; N, 7.65

(b) A suspension of **7** (0.1 mmol) and NaBH₄ (1.0 mmol) in methanol (2 mL) was stirred for 2 hrs at room temperature. Water was added, the methanol was evaporated *in vacuo* and the mixture was extracted with methylene chloride. The dried (MgSO₄) solution was evaporated to dryness and the crude reaction product was recrystallized from benzene-hexane to give *trans-8* in 72% yield, mp. 153-155°, identical with the described above.

cis-1-Benzyl-5-hydroxymethyl-3,4-diphenylimidazolidin-2-one (8b).- Obtained analogously with (a) in the above protocol in 39% yield from 0.1 mmol of *erythro-10* and purified by PTLC over silica gel [eluent hexane-diethyl ether (1:1 v/v)] before recrystallization. Mp. 130-132° (benzene-hexane).

Anal. Calcd. for C₂₃H₂₂N₂O₂: C, 77.07; H, 6.19; N, 7.82. Found: C, 77.14; H, 6.20; N, 7.78

Acknowledgment.- The authors thank the Bulgarian National Fund for Scientific Research for financial support.

REFERENCES

1. K. Kavrakova and M. J. Lyapova, *J. Chem. Res.(S)*, 186 (1993); Corrigendum, *J.Chem. Res.(S)*, 482 (1995).
2. A. Shanzer, L. Somekl and D. Butina, *J. Org. Chem.*, **44**, 3967 (1979).
3. *e.g.*: a) P. J. Dunn, R. Hner and H. Rapoport, *ibid.*, **55**, 5017 (1990); b) G. Cardillo, M. Orena, M. Penna, S.Sandri, and C. Tomasini, *Tetrahedron*, **47**, 2263 (1991); c) E. Pflammatter, D. Seebach, *Ann.*, 1323 (1991) and references cited therein.
4. T. Yamada, N. Isono, A. Inui, T. Miyazawa, S. Kuwata, and H. Watanabe, *Bull. Chem. Soc. Jpn*, **51**, 1878 (1978).

(Recived October 2, 1995; in revised form February 5, 1996)